

Triethylammonium hydrogen fumarate

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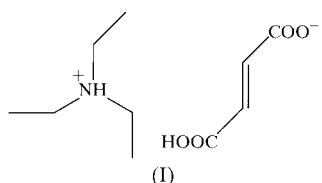
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In crystals of the title compound, the hydrogen fumarate anions form one-dimensional chains through an O—H \cdots O hydrogen bonding along the *c* and (*a*+*b*)/2 directions. There are three sites of the hydrogen fumarate, two of which have an inversion centre.

Comment

The crystal structures of the ammonium and isopropylammonium salts of the fumaric acid dianion were reported by Hosomi *et al.* (1998). In the title crystals, (I), *trans-cis* photoisomerization was observed, but no photodimerization occurred. There is a strong O5—H5—O7 hydrogen bond; the O5—H5 and O7—H5 distances are 1.19 (3) and 1.26 (3) Å, respectively.



Experimental

The crystals of (I) were grown from a diethyl ether/2-propanol solution.

Crystal data

$C_6H_{16}N^+\cdot C_4H_3O_4^-$	$Z = 4$
$M_r = 217.26$	$D_x = 1.166 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 13.409 (3) \text{ \AA}$	Cell parameters from 25 reflections
$b = 13.656 (3) \text{ \AA}$	$\theta = 14.4\text{--}15.0^\circ$
$c = 7.432 (2) \text{ \AA}$	$\mu = 0.089 \text{ mm}^{-1}$
$\alpha = 95.03 (3)^\circ$	$T = 295 (1) \text{ K}$
$\beta = 90.11 (2)^\circ$	Prism, colourless
$\gamma = 113.99 (1)^\circ$	$0.6 \times 0.6 \times 0.4 \text{ mm}$
$V = 1237.5 (5) \text{ \AA}^3$	

Data collection

Rigaku AFC-5 diffractometer	$h = 0 \rightarrow 17$
$\theta\text{--}2\theta$ scans	$k = -17 \rightarrow 17$
5071 measured reflections	$l = -9 \rightarrow 9$
4850 independent reflections	3 standard reflections
3040 reflections with $I > 2\sigma(I)$	every 100 reflections
$R_{\text{int}} = 0.016$	intensity decay: 4.4%
$\theta_{\text{max}} = 26.0^\circ$	

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0510P)^2 + 0.1858P]$
$R(F) = 0.045$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.117$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.02$	$\Delta\rho_{\text{max}} = 0.21 \text{ e \AA}^{-3}$
4850 reflections	$\Delta\rho_{\text{min}} = -0.12 \text{ e \AA}^{-3}$
424 parameters	Extinction correction: <i>SHELXL97</i> (Sheldrick, 1997)
All H-atom parameters refined	Extinction coefficient: 0.026 (2)

Table 1

Selected geometric parameters (Å, °).

O1—C11	1.255 (2)	O5—C15	1.288 (2)
O2—C11	1.235 (2)	O6—C15	1.217 (2)
O3—C14	1.293 (2)	O7—C17	1.268 (3)
O4—C14	1.208 (2)	O8—C17	1.226 (3)
O1—C11—O2	126.1 (2)	O5—C15—O6	125.6 (1)
O3—C14—O4	124.3 (2)	O7—C17—O8	125.7 (2)

Table 2

Hydrogen-bonding geometry (Å, °).

$D\text{--}H\cdots A$	$D\text{--}H$	$H\cdots A$	$D\cdots A$	$D\text{--}H\cdots A$
O3—H3 \cdots O1 ⁱ	1.08 (3)	1.40 (3)	2.481 (2)	174 (3)
O5—H5 \cdots O7	1.19 (3)	1.26 (3)	2.445 (2)	175 (4)
N9—H9 \cdots O2 ⁱⁱ	0.90 (2)	1.81 (2)	2.708 (2)	170 (2)
N10—H10 \cdots O8	0.87 (2)	1.92 (2)	2.781 (2)	173 (2)

Symmetry codes: (i) $x, y, z - 1$; (ii) $1 - x, -y, 2 - z$.

The C—H bond distances are in the range 0.89 (2)–1.05 (4) Å and the O—H distance not involving the H5 atom is 1.08 (3) Å.

Data collection: *MSC/AFC Diffractometer Control Software* (Molecular Structure Corporation, 1993); cell refinement: *MSC/AFC Diffractometer Control Software*; data reduction: *TEXSAN* (Molecular Structure Corporation, 1999); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); software used to prepare material for publication: *TEXSAN*.

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